

- 11) Falling sphere viscometer based on which law -----
 - a) Poiseuille law
 - b) Stoke law
 - c) Diffusion law
 - d) Ficks law
- 12) Chemical kinetic is the study of the -----
 - a) Rate of chemical Reaction
 - b) Particle size
 - c) Rheological Property
 - d) Interfacial tension
- 13) Kinematic viscosity is the ratio of dynamic viscosity to the
 - a) Density of fluid
 - b) Plastic viscosity
 - c) Volume of liquid to flow
 - d) Specific gravity of fluids.
- 14) Helium pycnometer is used to determine -----
 - a) Size
 - b) True density
 - c) Particle interaction
 - d) Surface area
- 15) For an ideal suspension, the sedimentation volume should be ---
 - a) 1
 - b) Less than 1
 - c) More than 1
 - d) Zero
- 16) The unit of viscosity is -----
 - a) N sec m^{-2}
 - b) $\text{N sec}^2 \text{ m}^2$
 - c) $\text{Ns ec}^{-1} \text{ m}^{-1}$
 - d) Newton
- 17) If the gold number is less than the protective action will be ----
 - a) More
 - b) Less
 - c) Half
 - d) Zero
- 18) When the size of particle is less than 1nm then it is called ----
 - a) Molecular dispersion
 - b) Colloidal dispersion
 - c) Suspension
 - d) Emulsion
- 19) In Negative thixotropy increase in viscosity curve moves ---
 - a) Upward
 - b) Right
 - c) Downward
 - d) None
- 20) Accelerated stability testing is done to -----
 - a) Predict Sheff of formulation
 - b) Predict dissociation constant
 - c) Predict diffusion constant
 - d) Determine activation energy

2. Solve **any two**.

**10x2
=20**

- a) Explain non Newtonian system in detail.
- b) Explain kinetics properties of colloids.
- c) Explain the methods for determining particle size and flow properties of powders.

3. Solve any seven.

**5x7
=35**

- a) Difference between lyophilic, lyophobic and association colloids.
- b) Enlist viscometer based on single point and multiple point determination. Explain working and principle of cub and bob viscometer.
- c) Define suspension. Explain interfacial properties of solids.
- d) Explain the chemical factors that affect the rate of reaction.
- e) Explain First Order reaction in detail.
- f) Add a note on formulation of flocculated and deflocculated suspensions.
- g) Explain physical stability of emulsion.
- h) Explain various methods for determining particle surface area.
- i) Explain stability of colloids.
